

DETERMINATION OF THE CONJUGACY CLASSES AND CHARACTER TABLE OF THE FULL NON-RIGID GROUP OF HEXACHLOROCYCLOPROPANE CHEMICAL COMPOUND VIA WREATH PRODUCT OF PAIR OF PERMUTATION GROUPS

Ben O. Johnson^{1*} and Auta T. Jonathan²

^{1&2}Department of Mathematics and Statistics,

Federal University, Wukari, Taraba State, Nigeria.

*Corresponding Author's Email: benjohnsonnig@yahoo.com

Cite this article:

Ben O. J., Auta T. J. (2024), Determination of the Conjugacy Classes and Character Table of the Full Non-Rigid Group of Hexachlorocyclopropane Chemical Compound Via Wreath Product of Pair of Permutation Groups. African Journal of Mathematics and Statistics Studies 7(1), 97-103. DOI: 10.52589/AJMSS-T9JSEVAG

Manuscript History

Received: 2 Nov 2023

Accepted: 17 Jan 2024

Published: 12 Feb 2024

Copyright © 2024 The Author(s). This is an Open Access article distributed under the terms of Creative Commons Attribution-NonCommercial-NoDerivatives 4.0 International (CC BY-NC-ND 4.0), which permits anyone to share, use, reproduce and redistribute in any medium, provided the original author and source are credited. **ABSTRACT**: The use of full non-rigid (f.NRG) molecules group theory to study the internal dynamics of molecular structures of chemical compounds is trending in the research space. In this paper, we use computational method to compute the group elements and group table of the Hexachlorocyclopropane molecular (algebraic) structure and thereafter determine the order and conjugacy classes of the group and finally the corresponding symmetry for each permutation group. We considered the point group of the compound which turns out to be isomorphic to the Wreath Products C₃wrC₂, where C_n denotes a cyclic group of order n. We found that the group has order 18 and 9 conjugacy classes. Investigating the solubility of the components of Hexachlorocyclopropane chemical compound can be considered for further study.

KEYWORDS: Non-Rigid Group, Hexachlorocyclopropane, Wreath Product, Character Table, Conjugacy Classes, Permutation Group



INTRODUCTION

The application of Group theory concepts to full non-rigid molecules is becoming increasingly relevant and is dense in the research space; see [10] and [12]. Symmetry as a concept is more of mathematics subject and not chemistry. However, the underlying mathematical theory for symmetry and group theory are of great importance in chemistry because they can be used to solve quite plenty of chemistry problems. For example it can be applied when it comes to the classification of the structures of molecules and crystals, understanding chemical bonding, predicting vibrational spectra and determining the optical activity of compounds.

Hexachlorocyclopropane exists as a chemical compound with molecular formula as C₃Cl₆. Full rigid molecules' symmetry properties are well known and so investigating them is a natural thing to do, though study in this research area is new in chemistry. [14] presented a number of papers on the determination of character tables of restricted f-NRG of some chemical molecules. For example, Computation of the r-NRG of the triple equivalent methyl rotation in pyramidal trimethylamine with inversion was done by [15]. The authors proved that the r-NRG of this molecule is a group has 648 elements, consists two subgroups of order 324 without inversion. We also have in [4] and [5] where computation of full non-rigid group of some other molecules were carried out. In the same vein, [1] computed full non-rigid group of tetra-ammine platinum (II), trimethylamine, tetraammine platinum (II), etc which are all examples of C2v and C4v point groups. Authors such as [2] and [7] proved that molecular structure tetra-amine platinum (II) is isomorphic to the Wreath Product of certain degree. [8] worked on hexamethylbenzene while [9] and [3] studied melamine.

MATERIALS AND METHODS

Let Ω be a nonempty set, a permutation of Ω is a bijection $\alpha: \Omega \to \Omega$. We denote the set of all permutations of Ω by S_{Ω} . When Ω is finite, that is, $\Omega = \{1, 2, ..., n\}$, we write S_n (the symmetric group of degree n) instead of S_{Ω} where $|S_n| = n!$ is the number of elements in S_n referred to as the order of the group S_n . Joseph-Louis Lagrange during his time observed permutation as arrangements, that is, as a list $i_1, i_2, ..., i_n$ with no repetition of any of the elements of Ω . Given an arrangement, $i_1, i_2, ..., i_n$, define a function $\alpha: \Omega \to \Omega$ by $\alpha(j) = i$ for all $j \in \Omega$. Thus, every rearrangement gives a bijection ([6]). The Wreath product of two permutation groups *C* by *D* denoted by W = C wr D is the semi-direct product of *P* (a set of mappings) by *D*, so that, $W = \{(f,d) | f \in P, d \in D\}$, with multiplication in *W* defined as $(f_1, d_1)(f_2, d_2) = ((f_1, f_2 d_1^{-1}) (d_1, d_2)) \forall f_1, f_2 \in P$ and $d_1, d_2 \in D$. *W* is a special form of permutation group. Understanding the nature of the Wreath products facilitates comprehension of certain types of subgroups of the symmetric groups.

In this paper, we carry out computation of Hexachlorocyclopropane. We start by first, specifying the algebraic structure of the non-rigid group of these molecules. Then, on the bases of the structure of the group, a useful programming language, namely, "Groups, Algorithms and Programming" (GAP) is applied and the character Tables is computed. The use of this computational group theory package (GAP) is capable of revealing many properties of the groups according to [11] and [13]. We determine the order of the f-NRG of



the group and it's conjugacy classes. Finally, we compute the character tables of the molecules with the aid of the computational group theory (GAP).

RESULTS AND DISCUSSIONS

The Hexachlorocyclopropane (C₃Cl₆) ion belongs to the Dihedral groups which are point groups that have n additional C₂ axes that stand perpendicular to the principal axis of the order 3. Since there are no other symmetry elements, the point group is of the type D_{3h} .

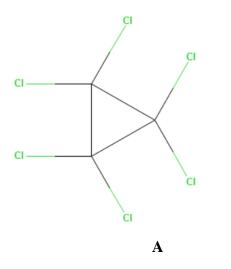


Figure 1: Hexachlorocyclopropane Structure.

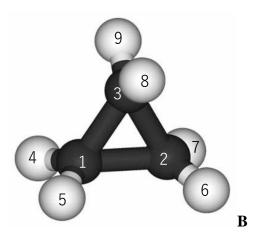


Figure 2: Hexachlorocyclopropane Geometry.

African Journal of Mathematics and Statistics Studies ISSN: 2689-5323 Volume 7, Issue 1, 2024 (pp. 97-103)



Where A = Chemical structure of Hexachlorocyclopropane

B = Geometry of Hexachlorocyclopropane

From figure 1 we have a C₃ principal axis, and three additional C₂ axes, but no other symmetry element. The C₃ axis stands perpendicular to the paper plane, and there are three C₂ axes in the paper plane. Therefore the symmetry of C₃Cl₆ has a trigonal bipyramidal shape. The C₃ axis goes through the centre of the atom of the C molecules, and the three C₂ axes go through the three equatorial Cl atoms. The C₃Cl₆ symmetry elements, namely C₃ and C₂ axes, are denoted by C₃ := $\langle (1,2,3) \rangle$ and C₂₁ := $\langle (4,5) \rangle$, C₂₂ := $\langle (6,7) \rangle$, C₂₃ := $\langle (8,9) \rangle$ respectively.

Therefore the full symmetry of Hexachlorocyclopropane as can be viewed clearly from figure 2, is: $G = (C_{21} \times C_{22} \times C_{23}) \rtimes C_3$, which we can write in terms of wreath product as $\coloneqq \mathbb{Z}_2 wr \mathbb{Z}_3$. We used GAP package to get the group as follows:

gap>

gap> C3 := Group((1,2,3));

Group([(1,2,3)])

gap> C2 := Group((4,5));

Group([(4,5)])

gap> G := WreathProduct(C3,C2);

Group([(1,2,3), (4,5,6), (1,4)(2,5)(3,6)])

gap> Order(G);

18

gap> CC := ConjugacyClasses(G);

[()^G, (4,5,6)^G, (4,6,5)^G, (1,2,3)(4,5,6)^G, (1,2,3)(4,6,5)^G, (1,3,2)(4,6,5)^G,

(1,4)(2,5)(3,6)^G, (1,4,2,5,3,6)^G, (1,4,3,6,2,5)^G]

gap>List(CC, x -> Order(Representative(x)));

[1, 3, 3, 3, 3, 3, 2, 6, 6]

gap> Display(CharacterTable(G));



S/N	Representatives	Size	Name
1.	(1)	1	1 <i>a</i>
2.	(4,5,6)	3	За
3.	(4,6,5)	3	3 <i>b</i>
4.	(1,2,3)(4,5,6)	3	3 <i>c</i>
5.	(1,2,3)(4,6,5)	3	3 <i>d</i>
6.	(1,3,2)(4,6,5)	3	3 <i>e</i>
7.	(1,4)(2,5)(3,6)	2	2a
8.	(1,4,2,5,3,6)	6	6a
9.	(1,4,3,6,2,5)	6	6 <i>b</i>

Table 2: Character Table for Hexachlorocyclopropane

	1a	3а	3 <i>b</i>	3 <i>c</i>	3d	3e	2a	6a	6 <i>b</i>
2 <i>P</i>	1a	3 <i>b</i>	3а	3 <i>e</i>	3 <i>d</i>	3 <i>c</i>	1a	3 <i>c</i>	3e
3 <i>P</i>	1a	1a	1a	1a	1a	1a	2a	2a	2a
5 <i>P</i>	1a	3b	3a	3e	3d	3c	2a	6b	6a
χ_1	1	1	1	1	1	1	1	1	1
χ_2	1	1	1	1	1	1	-1	-1	-1
χ3	1	Α	Ā	Ā	1	Α	-1	-A	$\overline{-A}$
χ_4	1	Ā	Α	Α	1	Ā	-1	$\overline{-A}$	-A
χ_5	1	Α	Ā	Ā	1	Α	1	Α	Ā
χ6	1	Ā	-1	Α	1	Ā	1	Ā	Α
χ_7	2	-1	Α	2	-1	2	0	0	0
χ_8	2	Ā	-A	В	-1	\overline{B}	0	0	0
χ9	2	-A	$\overline{-A}$	\overline{B}	-1	В	0	0	0

Where:

$$A = E(3)^2 = (-1 - Sqrt(-3))/2 = -1 - b3$$

$$B = 2 * E(3)^2 = -1 - Sqrt(-3) = -1 + i3$$

Note: Table 1 provides the Representatives of the conjugacy classes of the group while Table 2 gives the Character table. Across the top of the character table is a list of all the symmetry operations in the point group. Down the left side of the character table is a list of irreducible

African Journal of Mathematics and Statistics Studies ISSN: 2689-5323 Volume 7, Issue 1, 2024 (pp. 97-103)



representations. An irreducible representation describes how in this case the atoms changes when the different symmetry operations are performed. It could stay the same (1), become inverted (-1), or change completely (0). Representations with A and B labels indicate non-degeneracy. Just below the Table 2 are mathematical functions which can be used to assign orbitals to the irreducible representations.

CONCLUSIONS

In this work, the symmetry operations of the non-rigid Hexachlorocyclopropane were studied. The conjugacy classes and the irreducible character tables of the molecules were computed as seen in table 1 and 2 using computational group theory. We found that the group is isomorphic to the Wreath Products $\mathbb{Z}_2 wr\mathbb{Z}_3$ and has order 18 and 9 conjugacy classes. We used GAP 4.11.1 for our calculations. Clearly, the character table makes it easy to classify the wave functions, determine selection rules, and so on. Thus, symmetry can be applied in the determination of orbital overlap for molecular orbitals more quickly and easily. Note, only orbitals that have the same irreducible representation have the correct symmetry to overlap and form a molecular orbital. It is natural to follow this study with the investigation of the solubility of the components of Hexachlorocyclopropane chemical compound.

REFERENCES

- Ashrafi, A. R. and Hamadanian, M. (2003). The full non-rigid group theory for tetraaminoplatinum (II). Croat. Chem. Acta., 76: 299-303.
- Ashrafi, A. R. and Hamadanian, M. (2004). Group theory for tetraammine platinum (II) with C2v and C4v point group in the non-rigid system. J. Appl. Math. Comput., 14: 289-303.
- Ashrafi, A. R. and Hamadanian, M. (2005). Full non-rigid group theory and symmetry of melamine. J. Iran. Chem. Soc., 2: 135-139.
- Balasubramanian, K. (2004a). Nonrigid Group Theory, Tunneling Splittings, and Nuclear Spin Statistics of Water Pentamer: (H2O)5. J. Phys. Chem., 108: 5527-5536.
- Balasubramanian, K. (2004b). Group theoretical analysis of vibrational modes and rovibronic levels of extended aromatic C48N12 azafullerene. Chem. Phys. Letters, 391: 64-68.
- Burness, T. C. and Tong-Viet, H. P. (2016). Primitive permutation groups and derangements of prime power order. *Manuscripta Mathematics*. (150)3. Pp. 255–291.
- Darafsheh, M. R., Farjami, Y. and Ashrafi, A. R. (2005a). The non-rigid group of tetraamineplatinum (II) as a wreath product. Bull. Chem. Soc. Jpn., 78: 996-1000.
- Darafsheh, M. R., Ashrafi, A. R. and Darafsheh, A. (2005b). Computing the full non-rigid group of tetra-tertbutyltetrahedrane using wreath product. Int. J. Quantum Chem., 105: 485-492.
- Darafsheh, M. R., Ashrafi, A. R. and Darafsheh, A. (2006). The full non-rigid group of hexamethylbenzene using wreath product. Chem. Phys. Letters, 421: 566-570.
- Ezra, G. S. (1982). Symmetry Properties of Molecules, Lecture Note in Chemistry 28, Springer, pp. 655-661.
- Karimi, T. E., Moghadam, M. D. G., Farrokhi M. and Aghaei, M. (2011). The full non-rigid group theory for trimethylborane with C3v, C3h and CS point groups, J. Argent. Chem. Soc., pp. 67-73.

African Journal of Mathematics and Statistics Studies ISSN: 2689-5323 Volume 7, Issue 1, 2024 (pp. 97-103)



- Maruani, J. and Serre, J. (1983). Symmetries and Properties of Non-Rigid Molecules, (Eds.) Elsevier, Amsterdam, pp.88-92.
- Moghani, A., Naghdi, S. and Sorouhesh, M. R. (2010). The Fujita combinatorial enumeration for the non-rigid group of 2, 4- dimethylbenzene. J. Serb. Chem. Soc., 75(1): 91-99.
- Smeyers, Y. G. (1992). Introduction to group theory for non-rigid molecules. Adv. Quantum Chem., 24: 1-77.
- Smeyers YG, Villa M (2000). A study of the internal dynamics of trimethylamine by means of the non-rigid group theory. J. Math. Chem., 28: 377-388. The GAP Group (2008). GAP-Groups, Algorithms, and Programming, Version 4.4.12 (http://www.gapsystem.org)